

## Enhanced benthic response to upwelling of the Indonesian Throughflow onto the southern shelf of Timor-Leste, Timor Sea

Daniel M. Alongi,<sup>1</sup> Richard Brinkman,<sup>1</sup> Lindsay A. Trott,<sup>1</sup> Fernando da Silva,<sup>2</sup> Francisco Pereira,<sup>2</sup> and Tonny Wagey<sup>3</sup>

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[1] Benthic microbial metabolism and bacterial diagenetic pathways were measured along the southern shelf of Timor-Leste during an upwelling event in the winter SE monsoon season. Vertical profiles of water properties and bottom water nutrient concentrations, and operational ocean modeling showed subsurface upwelling from the Indonesian Throughflow (ITF) along the southern shelf west of longitude 126°25'E and surface upwelling at the far eastern end of the shelf. Warm surface waters above the halocline had salinities of 33.6 to 33.9 overlying cooler ITF water with salinities of 34.4 to 34.6. Beneath the zone of subsurface upwelling and stratification, sediment chlorophyll *a* (range: 2.8–4.4  $\mu\text{g g}^{-1}$ ) and phaeopigment (range: 4.5–7.0  $\mu\text{g g}^{-1}$ ) concentrations were sufficient to fuel very rapid rates of benthic oxygen consumption (range: 89.9–142.3  $\text{mmol m}^{-2} \text{day}^{-1}$ ) and dissolved inorganic carbon (DIC) release (range: 108.1–148.9  $\text{mmol m}^{-2} \text{day}^{-1}$ ) across the sediment-water interface, and DIC (range: 94.7–142.5  $\text{mmol m}^{-2} \text{day}^{-1}$ ) and  $\text{NH}_4^+$  (range: 13.3–19.9  $\text{mmol m}^{-2} \text{day}^{-1}$ ) production from incubated surface (0–10 cm) sediments. Molar ratios of DIC/ $\text{NH}_4^+$  production were lower (range: 6.6–7.7) in fine-grained sediments under the subsurface upwelling regime than in sandy, possibly scoured sediments under surface upwelling (range: 11.9–21.2) where there was no evidence of benthic enrichment. It is proposed that subsurface upwelling along the widest portions of the shelf stimulates phytoplankton production, leading to deposition of fresh phytodetritus that is rapidly decomposed on the seafloor. These zones of high biological activity may attract and support large populations of pelagic fish and cetaceans that have been caught for centuries along the south coast.

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### 1. Introduction

[2] The Timor Sea is an important oceanographic conduit linking the Pacific and Indian oceans. The main water mass linking both oceans is the Indonesian Throughflow (ITF), part of which transits through the Timor Sea. The ITF transports roughly 15 Sv from the Pacific to the Indian Ocean, forming the warm-water route for the global thermohaline circulation and providing a path for water to circumnavigate the global ocean [Gordon and Fine, 1996; Sprintall, 2009; Tillinger, 2011]. The ITF is thus closely linked to global climate via the El Niño/Southern Oscillation phenomenon and the Indian Ocean Dipole [Saji *et al.*, 1999; England

and Huang, 2005]. Due primarily to a pressure gradient induced by differences in sea level between the Pacific and Indian oceans, water from the North Pacific enters the Sulawesi Sea and continues through the Makassar Strait and, to a lesser extent, the Lifamatola Passage. The ITF then exits via Lombok Strait or circulates through the Banda and Flores seas and enters the Indian Ocean via Ombai Strait and Timor Passage [Atmadipoera *et al.*, 2009]. Recent estimates indicate that roughly half of ITF water outflows through the Timor Passage [Tillinger, 2011]. This relatively low-salinity water flows south of the island of Timor in a westerly direction within the upper 200 m at speeds up to 0.5  $\text{m s}^{-1}$  with a weak secondary flow near 1200 m depth [Molcard *et al.*, 1996]. The Timor Sea is under the influence of the NW monsoon from November to March when surface flow is northeasterly, and from May to September when the SE monsoon prevails, with surface flow reverting to the west from March to August [Cresswell *et al.*, 1993; Schiller *et al.*, 2010].

[3] Variations in the strength of the ITF have been linked inversely to variations in paleoproductivity in the Timor Sea [Müller and Opdyke, 2000; Holbourn *et al.*, 2005; Xu *et al.*, 2006]. Although coastal upwelling occurs at the edge of the northern Australian shelf [Furnas, 2007; McKinnon

<sup>1</sup>Australian Institute of Marine Science, Townsville, Queensland, Australia.

<sup>2</sup>Departamento de Geslav de Recursos Pesqueiros, Ministeris de Agricultura, Florestas e Pescas, Comoro, Dili, Timor Leste.

<sup>3</sup>Arafura and Timor Seas Ecosystem Action, Agency for Marine and Fisheries Research Jalan Mt. Haryono Kav, Jakarta, Indonesia.

Corresponding author: Daniel M. Alongi, Australian Institute of Marine Science, PMB 3, Townsville MC, Queensland 4810, Australia. (d.alongi@aims.gov.au)

*et al.*, 2011] and in the eastern Banda and northern Arafura seas [Gieskes *et al.*, 1990], it is not known what effect, if any, the ITF has on productivity in the northern Timor Sea in close proximity of the island of Timor. The region is ecologically and physically diverse, yet little scientific information is available. The narrow southern shelf of Timor was once sampled extensively in 1900 by the Siboga Expedition [Weber, 1902], which identified areas of fine sediment along most of the Timorese coast, with coarser sand and coral deposits east of longitude 127°.

[4] To address the paucity of information and as part of a larger project on the Timor and Arafura seas (<http://atsea-program.org>), we conducted a benthic survey along the south coast of Timor-Leste during the SE monsoon season to measure rates and pathways of sediment metabolism in relation to sediment grain size, nutrient concentrations, and the origin of sedimentary organic matter. As detailed here, we found very rapid rates of metabolism along sections of the shelf edge that appear to be in response to upwelling of water from the ITF.

## 2. Methods

### 2.1. Study Site and Field Sampling

[5] The continental shelf of southern Timor-Leste (Figure 1) is a narrow, active margin that is part of the Banda Arc where isostatic rebound of previously subducted Australian crust has resulted in regional uplift [Harris, 1991; Richardson and Blundell, 1996]. This portion of the Banda Arc is a nonvolcanic collision zone, but uplift is active today as evidenced by the widespread occurrence of raised Pliocene to Holocene coral reef terraces hundreds of meters thick along the coast [Chappell and Veeh, 1978].

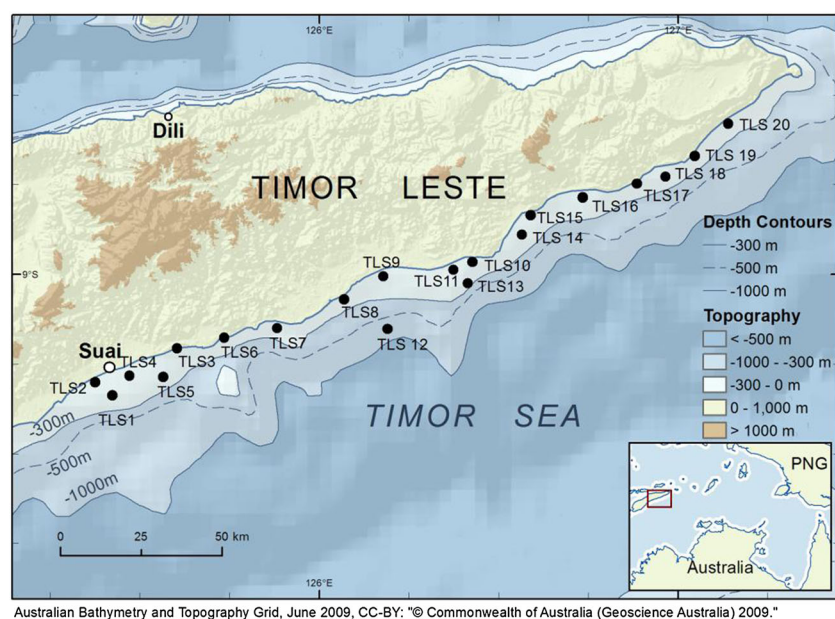
[6] Timor-Leste experiences a dry season from July to October and a wet season from November to March with an average annual rainfall of 1500 to 2000 mm along the

mountainous spine of the island and along the southern coast [Durand, 2002]. The south coast contains more than 20 small rivers that drain rapidly eroding, short, steep catchments. Rapid uplift, heavy rainfall, rapid weathering, and extensive deforestation result in an average sediment load of 60 Mt yr<sup>-1</sup> [Milliman *et al.*, 1999]. The narrow shelf (defined here as the ≤300 m contour) along the south coast is 11 km wide at its widest point near Suai, but shelf width is ≤2 km at several capes along the coast. Total shelf area is approximately 1580 km<sup>2</sup> along a coastline length of 285 km for an average shelf width of about 5.5 km.

[7] Twenty stations were visited from 10 to 18 July 2011 on the RV *Solander*. These stations covered the entire length of the south coast (Figure 1). At each station, conductivity-temperature-depth (CTD) and Niskin casts were taken for measurement of water-column structure and bottom water nutrients, and benthic grab or boxcorer samples were taken for measurements of sediment characteristics and microbial metabolism.

### 2.2. Water-Column Measurements and Oceanic Modeling

[8] CTD casts at each station were made using a Seabird SBE911Plus Livewire CTD (Sea-Bird Electronics, Inc., Bellevue, Wash.) fitted with a LICOR Photosynthetically Active Radiation (PAR) sensor (LICOR, Lincoln, Nebr.), a Seabird SBE43 oxygen sensor, and a Chelsea chlorophyll fluorometer (Chelsea Technologies, West Molesey, Surrey, UK). A separate cast was taken using a 10-L Niskin bottle to obtain water samples for measurement of water-column nutrients 1 m above the seafloor. Water samples were collected from Niskin bottles using acid-washed syringes and filtered through a 0.45 μm filter cartridge (Sartorius Minisart) into acid-washed screw-cap plastic test tubes and stored frozen (-18°C) until later analysis ashore. DOC samples were acidified with 100 μl analytical reagent-grade



**Figure 1.** Chart of the 20 benthic stations along the southern shelf of Timor-Leste sampled during July 2011. © Commonwealth of Australia (Geoscience Australia), 2009.

HCl and stored at 4°C until analysis. Inorganic dissolved nutrient concentrations were determined by standard wet chemical methods [Ryle *et al.*, 1981] implemented on a segmented flow analyzer [Bran and Luebbe, 1997].

[9] Thicknesses of the mixed layer, the isothermal layer, and the top of the thermocline were estimated using the gradient analysis criteria of Lucas and Lindstrom [1991]. A 5 m starting depth was used to avoid misclassifications due to near-surface disturbance. A density gradient of 0.01 kg m<sup>-1</sup> was used to determine the mixed-layer depth. To determine the thickness of the isothermal layer and the top of the thermocline, we used critical temperature gradients of 0.025°C m<sup>-1</sup> and 0.05°C m<sup>-1</sup>. CTD profiles at each station were searched downward until each of the criteria was met between at least three consecutive 1 m depths. Layer depth was assigned the shallowest value.

[10] Three-dimensional descriptions of potential temperature during the period of this study were obtained from the Australian Bureau of Meteorology OceanMAPS forecasting system, based on an operational implementation of the Ocean Forecast Australia Model (OFAM) [Schiller *et al.*, 2008]. OFAM is based on version 4.0 of the Modular Ocean Model [Griffies *et al.*, 2004], with a resolution grading from 2° in the North Atlantic to 1/10° in the Asian–Australian region from 90°E to 180°E and from 16°N to 75°S. The model grid contains 47 levels in the vertical, 35 of which are in the top 1000 m, with 10 m resolution near the surface. The model uses hybrid mixed-layer representation; horizontal viscosity is resolution and state dependent based on the Smagorinsky scheme, and due to the model's variable grid size, anisotropic options have been chosen for this parameterization [Schiller *et al.*, 2008]. The performance of the OFAM model in the region covering the shelf of southern Timor-Leste has been evaluated against observations of potential temperature and velocity from moorings within the straits of Timor and Ombai, and the model agrees reasonably well with these observations [Schiller *et al.*, 2010].

### 2.3. Bulk Sediment Measurements

[11] Sediment samples were taken from a 0.2 m<sup>2</sup> Smith-McIntyre grab with rubber-sealed lids. Samples were taken only from sediments having intact surface sediment structures such as animal tracks, tubes, and burrow openings. Surface (0–5 cm) samples were taken for measurement of total carbon (TC), total organic carbon (TOC), and total nitrogen (TN) content and for measurement of <sup>13</sup>δC and <sup>15</sup>δN content and for chlorophyll *a* and phaeopigments. A large sample (>100 g) was also taken for sediment granulometry. Grain size was determined on an automated particle size counter and sediment type classified based on the definitions in Folk (1974). Samples for TC, TOC, and TN were frozen, wet-weighted, and dry-weighted to determine water content and porosity before and after freeze-drying and ground to a fine powder for determination of TC and TN on a Perkin-Elmer 2400 CHNS/O Series II Analyzer (Perkin-Elmer, Waltham, MA, USA) and for TOC on a Shimadzu TOC Analyzer with solid sampler (Shimadzu Corp., Tokyo, Japan). Analytical performance was monitored with standard reference materials NBS 1646 (estuarine sediment) from National Institute of Standards Technology and BCSS-1 (marine sediment) obtained from the National Research Council of

Canada. Values were always within the certified range. Detection limits for solid-phase metals were 5 μg g<sup>-1</sup> sediment dry weight on the Thermo IRIS AAS. Total inorganic carbon was assumed as CaCO<sub>3</sub> and was determined by difference between the TC and TOC concentrations.

[12] Samples for benthic pigments were taken from surface (0–2 cm) sediments using a 5 ml syringe with the needle end cut off. Chlorophyll *a* and phaeopigments were extracted using a 1:1 (v/v) chloroform-methanol solution [Wood, 1985]. Absorbance was measured at 665 nm on a Varian spectrometer. The spectrophotometric equations of Lorenzen [1967] were modified to account for the use of a different extractant.

[13] Samples for stable isotopes were collected using rubber gloves and a clean metal spatula to avoid contamination. Samples were placed in clean glass vials and frozen until return to the lab. Samples were then freeze-dried and stored in airtight desiccators with silica gel. Subsamples (1 g) were transferred to 15 ml centrifuge tubes and 7 ml cold 10% HCl was added, 0.5 ml at a time to avoid overflow. Samples were left to react overnight; drops of cold acid were then added if necessary to complete the dissolution of carbonate. Once completely acidified, samples were washed with cold distilled water twice and oven-dried (60°C for 1–2 days). Samples were left in desiccators until analysis. Each sample was weighed and packed tightly in individual tin SerCon capsules (5 mm width, 8 mm height) and compressed, ensuring no air was left in the capsules. Samples were analyzed for N and C content (%), <sup>δ15</sup>N, and <sup>δ13</sup>C using an Automated Nitrogen Carbon Analyzer-Mass Spectrometer consisting of a 20/20 mass spectrometer connected with an ANCA-S1 preparation system (Europa Scientific Ltd., Crewe, UK). Results were normalized according to Paul *et al.* [2007].

[14] A simple two end-member mixing equation [Tyson, 1997, p. 396] was used to estimate the fraction of <sup>δ13</sup>C derived from terrestrial and marine sources. The terrestrial (–27‰) and marine (–19‰) end-members of Aller *et al.*, [2008] from the Gulf of Papua were used.

### 2.4. Fluxes Across the Sediment-Water Interface

[15] Solute fluxes across the sediment-water interface at each station were measured using three to six opaque chambers (volume: 1 L; area: 82 cm<sup>2</sup>; height: 20.5 cm) from which DIC samples were taken at 1 h intervals for 4–6 h. Each chamber was gently placed into the surface sediment of minimally disturbed grab samples. Each chamber was then withdrawn after a fitted plastic bottom lined with soft rubber was placed and fit to the chamber bottom by hand. The outside of each chamber was washed, and the entire set of chambers was then incubated under shade in a running seawater bath to maintain in situ seawater temperature. Each chamber had a propeller-electric motor unit placed within the top opening of the chamber. Three sampling ports were located on opposite sides of the chamber: (1) one port was fitted with tubing that opened to allow replacement water to enter [Alongi *et al.*, 2007]; (2) the second port was fitted with an O<sub>2</sub> probe (TPS Model WP-82 DO meters) to measure dissolved oxygen flux; and (3) the third port was fitted with plastic tubing to draw off 10-ml samples for DIC and dissolved inorganic nutrients (DIN). Samples were filtered (0.45 μm Minisart filters) and kept cool and dark for DIC, and frozen for DIN. Concentrations of DIC were determined

using the procedure of *Hall and Aller* [1992], and DIN concentrations were determined as described in section 2.2.

**2.5. Incubation Experiments: DIC and NH<sub>4</sub><sup>+</sup> Production and Metal Reduction in Sediments**

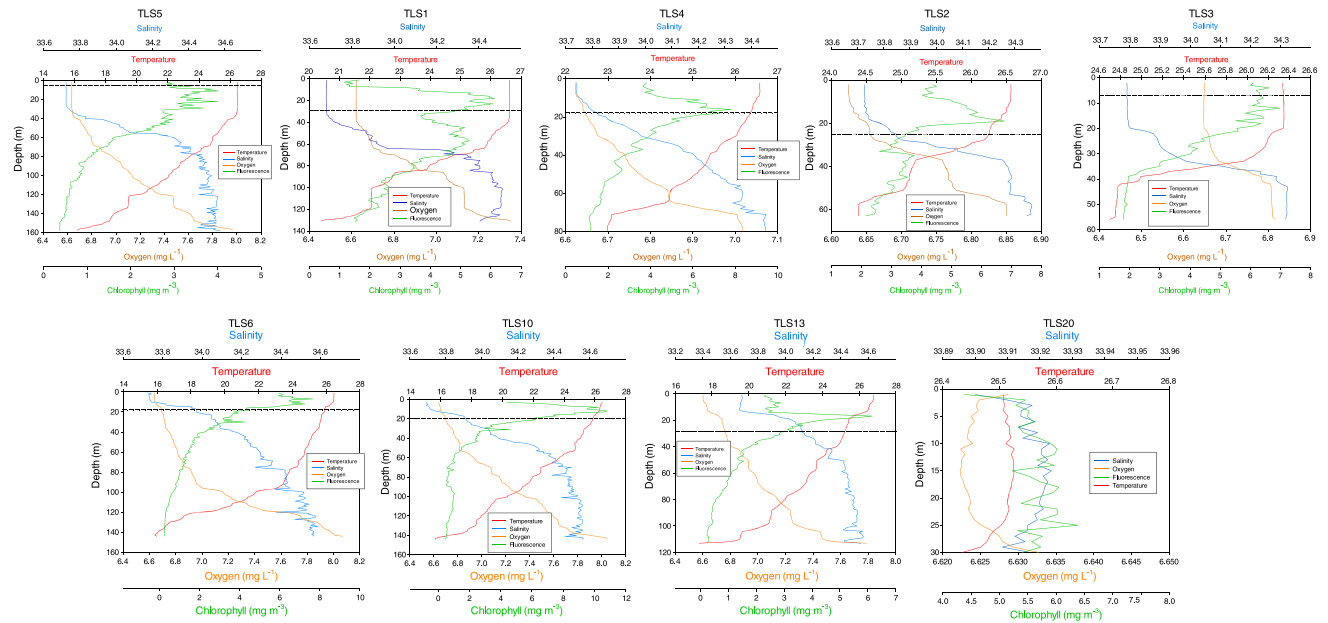
[16] Rates of net release of dissolved Fe, Mn, DIC, and NH<sub>4</sub><sup>+</sup> from sediments at each site were measured by using

a sediment incubation technique [*Alongi et al.*, 2007]. At each station, sediment slices were taken over the 0–10 cm depth interval by sinking opaque glass jars into the sediment. Further sediment was added so no airspace was present; then each jar was capped and sealed with electrical tape. The jars were then incubated at in situ temperature for 6–12 d. At the end of each incubation, 30–40 ml homogenized sediment

**Table 1.** Mean Water-Column Characteristics From CTD Casts Taken at Each of the Stations Along the South Timorese Coast<sup>a</sup>

Station	Latitude (S)	Longitude (E)	1% PAR (m)	WD (m)	MLD (m)	ITL (m)	TT (m)	SST/SBT (°C)	SSS/SBS	SSDO/SBDO (mg L <sup>-1</sup> )
TLS1	9°20.172'	125°25.36'	27	133	39	39	42	26.59/20.18	33.68/34.55	6.62/7.37
TLS2	9°17.971'	125°22.52'	24	52	10	10	31	26.57/24.40	33.73/34.34	6.62/6.84
TLS3	9°12.335'	125°36.20'	7	50	19	19	19	26.34/24.71	33.79/34.33	6.65/6.81
TLS4	9°16.880'	125°28.26'	21	81	9	6	9	26.58/22.98	33.74/34.45	6.62/7.02
TLS5	9°17.143'	125°33.88'	5	157	37	36	38	26.49/16.17	33.72/34.59	6.63/7.96
TLS6	9°10.493'	125°44.09'	19	144	10	10	10	26.46/16.12	33.73/34.56	6.63/8.25
TLS7	9°08.947'	125°52.91'	10	88	11	11	11	26.10/23.56	33.86/34.42	6.67/6.95
TLS8	9°04.172'	126°04.11'	14	105	6	6	9	26.46/20.63	33.66/34.53	6.64/7.32
TLS9	9°00.288'	126°10.74'	15	60	33	32	33	26.20/24.72	33.76/34.27	6.66/6.82
TLS10	8°57.908'	125°25.61'	20	148	10	6	35	26.45/15.66	33.71/34.56	6.64/8.04
TLS11	8°59.201'	126°22.44'	9	39	9	8	9	26.45/24.91	33.77/34.26	6.63/6.79
TLS13	9°01.496'	126°24.82'	23	120	15	14	16	26.78/18.56	33.73/34.45	6.60/7.60
TLS14	8°53.338'	126°33.91'	2	94	30	29	30	26.77/22.21	33.77/33.46	6.60/7.11
TLS15	8°50.048'	126°35.30'	5	32	30	27	28	26.46/26.14	33.60/33.86	6.64/6.67
TLS16	8°47.063'	126°44.07'	13	39	32	17	28	26.45/26.31	33.81/33.88	6.63/6.64
TLS17	8°44.790'	126°53.09'	17	62	62	—	—	26.44/26.32	33.81/33.89	6.64/6.64
TLS18	8°43.626'	126°57.92'	17	93	61	36	37	26.36/24.11	33.92/34.39	6.64/6.88
TLS19	8°40.126'	127°02.74'	14	70	70	—	—	26.40/23.25	33.87/34.46	6.63/6.98
TLS20	8°34.692'	127°08.38'	12	31	31	—	—	26.27/25.16	33.72/34.11	6.66/6.77

<sup>a</sup>ITL, thickness of isothermal layer; MLD, mixed layer depth; 1% PAR, depth of 1% incident light; SBDO, sea bottom dissolved oxygen; SBS, sea bottom salinity; SBT, sea bottom temperature; SSDO, sea surface dissolved oxygen; SSS, sea surface salinity; SST, sea surface temperature; TT, top of thermocline; WD, water depth.



**Figure 2.** Vertical structure of water properties and the 1% light level (horizontal line in each panel) across the widest section of the shelf (top row, from shelf-edge Sta. TLS5 to inshore Sta. TLS3), upwelling at the narrowest section (Sta. TLS6, bottom left) and further east (Stas. TLS10 and TLS13, bottom middle panels) along the shelf. East of longitude 126°25', inshore waters were well mixed (e.g., Sta. TLS20, bottom right).

from each jar was placed into a 50 ml centrifuge tube and centrifuged, after which the supernatant was filtered for DIC,  $\text{SO}_4$ ,  $\text{NH}_4^+$ , Ca, Mg, and Cl analysis. A 1 N KCl solution was then added to each centrifuge tube, mixed into each sample, and after 2–3 h incubation centrifuged again to obtain total extractable  $\text{NH}_4^+$ , which was determined using automated techniques described in section 2.2.

[17] To obtain preincubation concentrations of all solutes, another set of sediment samples from the same five to eight depth intervals was taken concurrently and centrifuged immediately to obtain pore water as described earlier. Each incubated jar sample was used for determination of both DIC and  $\text{NH}_4^+$  reaction rates. DIC was processed as described in section 2.4.

[18] For Fe and Mn reduction, two different sets of jar samples were processed as described earlier. Pore water  $\text{SO}_4^{2-}$  concentration in jar sediments was increased by  $10 \text{ mmol L}^{-1}$  to avoid depletion during incubation and mixed thoroughly under a constant  $\text{N}_2$  stream. The homogeneous sediment mixtures from each depth interval were transferred into eight 20 ml glass scintillation vials, which were capped with no headspace, taped to prevent oxygen intrusion, and incubated in the dark as close as possible to in situ temperature. Two vials were sacrificed at 6 to 12 day intervals for determination of pore water  $\text{Fe}^{2+}$  concentrations. Pore water was extracted from the jars by centrifuging at 2500 rpm for 15 min, returned to the glove bag, and filtered through Whatman GF/F filters. Samples were then acidified ( $20 \mu\text{l}$  of  $0.5 \text{ mol L}^{-1}$  HCl per 1 ml) and stored at  $5^\circ\text{C}$  in polyethylene vials until analysis by the spectrophotometric ferrozine technique [Stokey, 1970], in which a  $50 \mu\text{l}$  sample is transferred to 2 ml of 0.02% ferrozine in  $50 \text{ mmol L}^{-1}$  4-(2-hydroxyethyl)-1-piperazine ethanesulfonic acid buffer, pH 7.

[19] The sediment remaining after pore water extraction was homogenized under  $\text{N}_2$  for reactive Fe(II) extraction by a modified version of the HCl technique of Lovley and Phillips [1987]. In brief, 100–300 mg subsamples were extracted in 5 ml of  $0.5 \text{ mol L}^{-1}$  HCl for 15 min on a shaking platform at  $25^\circ\text{C}$  and centrifuged (5000 rpm for 10 min). The supernatant was GF/F filtered and stored at  $5^\circ\text{C}$  until analysis as described earlier for  $\text{Fe}^{2+}$ . Total dissolved Mn in both extracts was analyzed on the inductively coupled plasma-atomic absorption spectrometer. It was not possible to determine Mn oxidation state due to interference from extractable  $\text{Fe}^{2+}$ .

[20] Reaction rates were calculated from a linear fit of concentration changes in the time series of samples. Metal reduction rates (corrected for compaction during pore water extraction) were determined as reactive Fe(II) and Mn accumulation assuming limited precipitation into nonacid extractable phases [Canfield *et al.*, 1993]. Reactive amorphous oxyhydroxide concentrations were operationally defined as the difference between total Fe and Mn content measured from acidified samples on the ICP-AES.

## 2.6. Sulfate Reduction

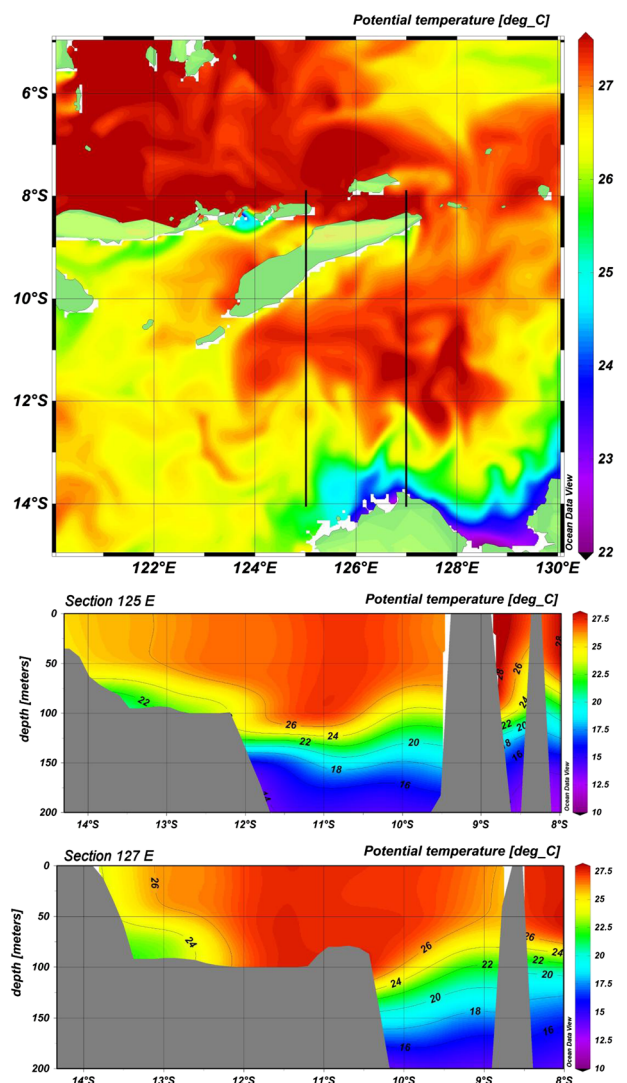
[21] Sulfate reduction was measured separately in triplicate 2.7 cm diameter plastic cores taken from surface to bottom of each grab sample to a maximum depth of 20 cm. Cores were injected at 1 cm intervals with carrier-free  $^{35}\text{S}$  [Fossing and Jorgensen, 1989], incubated in time series at 20 min and at 1 h, 3 h, 6 h, and 9 h at stations (Stas.) TLS1,

TLS4 and TLS20, and for 6–9 h at the other stations. Incubations were terminated by immediately fixing sediments in 20% zinc acetate. Samples were then frozen until a two-step distillation procedure was used to determine the fraction of reduced radiolabel shunted into the acid-volatile sulfide and chromium-reducible sulfur pools.

## 3. Results

### 3.1. Water-Column Structure and Nutrients

[22] Timor shelf stations were between 32 and 157 m in depth and, except for Stas. TLS17, TLS19, and TLS20, the water-column was stratified with the mixed-layer depth varying from 6 to 70 m (Table 1). Except for Stas. TLS2,



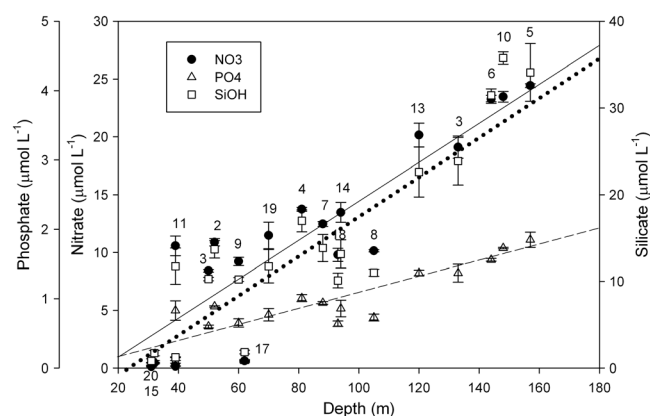
**Figure 3.** Synoptic view of surface temperature from OceanMAPS model on 17 July 2011 (top) with vertical sections of potential temperature along the longitude  $125^\circ$  (middle) and  $127^\circ$  (bottom) meridians, showing only surface 200 m. Thick lines in top panel show locations of meridional sections. Note change in color scaling between synoptic and sections plots.

TLS10, and TLS18, estimated depth of the mixed layer was shallower and within a few meters of the estimates of the isothermal layer and the top of the thermocline (Table 1) and, at most stations, shallower than the top of the halocline. Surface layers were well mixed at most stations with high turbidity, resulting in the 1% isolume occurring as shallow as less than 10 m at Stas. TLS3, TLS5, TLS7, TLS11, TLS14, and TLS15 (Table 1).

[23] Across the widest section of the shelf (Figure 2, top, from left to right), subsurface intrusions of cool (<20.2°C), nutrient-rich water were evident at Stas. TLS5, TLS4, and TLS1; further landward, the shelf was still stratified within 2 km from small river catchments (Stas. TLS2 and TLS3), but bottom temperatures were warm (>24°C) and dissolved nutrient concentrations low. Vertical structure of water properties indicated upwelling at a very narrow section of the shelf (Sta. TLS6, Figure 2, bottom left), as well as along another wide section of the shelf edge further east (Stas. TLS10 and TLS13, Figure 2, bottom two middle panels). At most of the upwelling stations, there was a well-defined chlorophyll maximum just above the 1% light level (e.g., Stas. TLS1, TLS4, TLS6, TLS10, TLS13, Figure 2). As noted earlier, waters were well mixed at Stas. TLS17, TLS19, and TLS20 (Figure 2, bottom right).

[24] Surface and subsurface intrusions of ITF water onto the southern Timorese shelf was confirmed by a synoptic view of surface temperature (Figure 3, top) on 17 July 2011 from the OceanMAPS model. The 127°E meridional section (Figure 3, bottom) shows a persistent surface signature of upwelling along the eastern end of the southern coast, whereas the 125°E meridional section (Figure 3, middle) confirms subsurface intrusions beneath warm surface waters.

[25] Bottom water phosphate ( $r=+0.887$ ), nitrate ( $r=+0.899$ ), and silicate ( $r=+0.902$ ) concentrations correlated positively ( $P < 0.0001$ ) with water depth (Figure 4).



**Figure 4.** Bottom water concentrations ( $\mu\text{mol L}^{-1}$ ) of nitrate, phosphate, and silicate in relation to water depth along southern Timor-Leste. Station numbers are listed above each set of values for a given water depth. Error bars are  $\pm 1$  standard deviation. All regressions are significant ( $P < 0.001$ ). Nitrate (solid line) =  $-2.786 + 0.172x$ ,  $R^2 = 0.809$ ,  $F_{1,18} = 71.866$ ; phosphate (dashed line) =  $-0.142 + 0.0118x$ ,  $R^2 = 0.786$ ,  $F_{1,18} = 62.428$ ; silicate (dotted line) =  $-5.193 + 0.235x$ ,  $R^2 = 0.813$ ,  $F_{1,18} = 73.765$ .

**Table 2.** Mean Sediment Granulometry and Composition, TOC and TN Concentrations, Stable Isotope Values, Percentages of Terrestrially Derived Organic Carbon, and Surface Benthic Chlorophyll  $\alpha$  and Phaeopigment Concentrations at Stations Along the South Timorese Coast (values are mean  $\pm 1$  standard deviation)<sup>a</sup>

Station	Grain Size ( $\mu\text{m}$ )	Silt (% WW)	Clay (% WW)	Sand (% WW)	TOC (% sed DW)	TN (% sed DW)	$\delta^{15}\text{N}$ (‰)	$\delta^{13}\text{C}_{\text{ORG}}$ (‰)	OC <sub>TERR</sub> (%)	Chlorophyll $\alpha$ ( $\mu\text{g g}^{-1}$ )	Phaeopigment ( $\mu\text{g g}^{-1}$ )
TLS1	7.6 (vfs)	71.7	23.5	4.8	0.65 $\pm$ 0.02	0.05 $\pm$ 0.01	2.86 $\pm$ 0.25	-24.38 $\pm$ 0.10	48	2.8 $\pm$ 0.3	4.5 $\pm$ 1.0
TLS2	7.6 (vfs)	73.6	22.1	4.3	0.50 $\pm$ 0.07	0.04 $\pm$ 0.01	2.64 $\pm$ 0.18	-24.38 $\pm$ 0.04	48	0.8 $\pm$ 0.6	0.2 $\pm$ 0.0
TLS3	8.4 (fs)	73.0	21.9	5.1	0.65 $\pm$ 0.02	0.05 $\pm$ 0.01	2.83 $\pm$ 0.42	-23.97 $\pm$ 0.09	43	0.5 $\pm$ 0.3	0.1 $\pm$ 0.1
TLS4	16.3 (ms)	74.5	23.2	2.3	0.60 $\pm$ 0.01	0.05 $\pm$ 0.01	2.68 $\pm$ 0.19	-24.64 $\pm$ 0.08	51	1.0 $\pm$ 0.5	2.2 $\pm$ 0.7
TLS5	64.9 (vfsd)	40.3	8.9	50.8	0.49 $\pm$ 0.01	0.04 $\pm$ 0.01	3.18 $\pm$ 0.24	-23.61 $\pm$ 0.12	38	4.4 $\pm$ 0.2	5.5 $\pm$ 1.5
TLS6	7.7 (vfs)	72.2	26.7	1.1	0.72 $\pm$ 0.07	0.06 $\pm$ 0.01	2.66 $\pm$ 0.33	-24.64 $\pm$ 0.24	51	4.0 $\pm$ 0.3	6.4 $\pm$ 2.3
TLS7	14.5 (fs)	74.0	22.0	4.0	0.63 $\pm$ 0.03	0.05 $\pm$ 0.01	2.76 $\pm$ 0.18	-23.74 $\pm$ 0.15	40	0.8 $\pm$ 0.6	1.4 $\pm$ 1.1
TLS8	17.2 (ms)	74.1	17.4	8.5	0.56 $\pm$ 0.04	0.03 $\pm$ 0.0	2.20 $\pm$ 0.21	-25.48 $\pm$ 0.49	62	0.4 $\pm$ 0.2	0.0 $\pm$ 0.0
TLS9	7.7 (vfs)	73.4	26.4	0.2	0.75 $\pm$ 0.06	0.05 $\pm$ 0.0	2.67 $\pm$ 0.24	-24.61 $\pm$ 0.18	51	0.3 $\pm$ 0.2	0.0 $\pm$ 0.0
TLS10	83.7 (vfsd)	36.5	9.1	54.4	0.69 $\pm$ 0.03	0.06 $\pm$ 0.01	3.51 $\pm$ 0.48	-22.44 $\pm$ 0.07	24	2.9 $\pm$ 0.2	7.0 $\pm$ 2.2
TLS11	10.1 (fs)	77.6	18.4	4.0	0.78 $\pm$ 0.02	0.06 $\pm$ 0.01	2.87 $\pm$ 0.91	-23.56 $\pm$ 0.11	38	0.6 $\pm$ 0.3	0.3 $\pm$ 0.1
TLS13	431.8 (msd)	19.2	6.2	74.6	0.28 $\pm$ 0.01	0.03 $\pm$ 0.0	4.13 $\pm$ 0.30	-22.27 $\pm$ 0.06	22	3.1 $\pm$ 0.3	6.5 $\pm$ 1.9
TLS14	9.9 (fs)	79.9	18.2	1.9	0.77 $\pm$ 0.04	0.07 $\pm$ 0.01	3.11 $\pm$ 0.43	-23.65 $\pm$ 0.22	39	0.7 $\pm$ 0.3	0.3 $\pm$ 0.1
TLS15	9.8 (fs)	74.4	17.3	8.3	0.83 $\pm$ 0.05	0.07 $\pm$ 0.01	2.25 $\pm$ 0.42	-25.14 $\pm$ 0.06	57	0.0 $\pm$ 0.0	0.0 $\pm$ 0.0
TLS16	14.3 (fs)	74.2	17.9	7.9	0.73 $\pm$ 0.03	0.06 $\pm$ 0.02	2.43 $\pm$ 0.11	-25.67 $\pm$ 0.26	64	0.1 $\pm$ 0.1	0.0 $\pm$ 0.0
TLS17	38.7 (cs)	75.1	13.4	11.5	0.80 $\pm$ 0.04	0.07 $\pm$ 0.01	2.42 $\pm$ 0.27	-25.52 $\pm$ 0.16	62	0.1 $\pm$ 0.1	0.0 $\pm$ 0.0
TLS18	14.4 (fs)	70.1	21.0	8.9	0.60 $\pm$ 0.01	0.06 $\pm$ 0.01	2.72 $\pm$ 0.56	-24.74 $\pm$ 0.16	52	0.3 $\pm$ 0.1	0.0 $\pm$ 0.0
TLS19	158.9 (fsd)	42.7	8.8	48.5	0.53 $\pm$ 0.02	0.06 $\pm$ 0.01	2.51 $\pm$ 0.91	-24.38 $\pm$ 0.31	48	0.4 $\pm$ 0.1	0.0 $\pm$ 0.0
TLS20	109.6 (vfsd)	45.6	7.3	47.1	2.00 $\pm$ 0.66	0.10 $\pm$ 0.01	1.62 $\pm$ 0.16	-28.65 $\pm$ 0.31	100	0.0 $\pm$ 0.0	0.0 $\pm$ 0.0

<sup>a</sup>cs, coarse silt; DW, dry weight; fs, fine silt; fsd, fine silt; msd, medium sand; vfs, very fine silt; vfsd, very fine sand; WW, wet weight.

### 3.2. Sediment Granulometry, Nutrients, and Plant Pigments

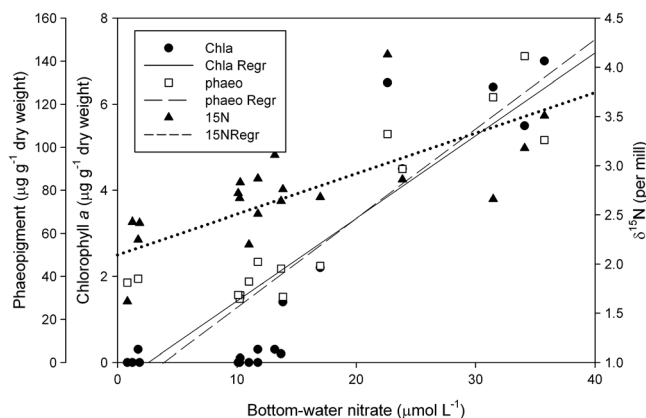
[26] Surface sediments at Stas. TLS5, TLS10, TLS13, TLS19, and TLS20 were composed of very fine to medium quartz sand; the remaining stations were composed of very fine to coarse silt, with amounts of clay ranging from 13.4% to 26.7% sediment wet weight (Table 2). Concentrations of TOC and TN along the shelf ranged from 0.28% to 2.0% and from 0.03% to 0.10%, respectively (Table 2), with highest values recorded at Sta. TLS20 where large amounts of plant detritus were found. Both TOC and TN correlated inversely, but weakly (both  $r = -0.435$ ;  $P < 0.0626$ ), with water depth.

[27] Sediment  $\delta^{13}\text{C}_{\text{ORG}}$  and  $\delta^{15}\text{N}$  values ranged from  $-28.65$  to  $-22.44$  and from  $1.62$  to  $4.13$ , respectively, and correlated with water depth ( $r = +0.523$ ,  $P < 0.0215$ ; and  $r = +0.593$ ,  $P < 0.00749$ , respectively). Lowest stable isotope values were recorded at Sta. TLS20. The percentage of sediment organic carbon of terrestrial origin varied from 24% to 100% (Table 2), with highest percentages recorded closest to the coast, as reflected in the inverse correlation with water depth ( $r = -0.527$ ;  $P < 0.0205$ ). Molar C:N ratios varied from 6.6 to 21.2, with the highest ratio recorded at Sta. TLS20; C:N ratios did not correlate significantly ( $P > 0.05$ ) with water depth ( $r = +0.0669$ ).

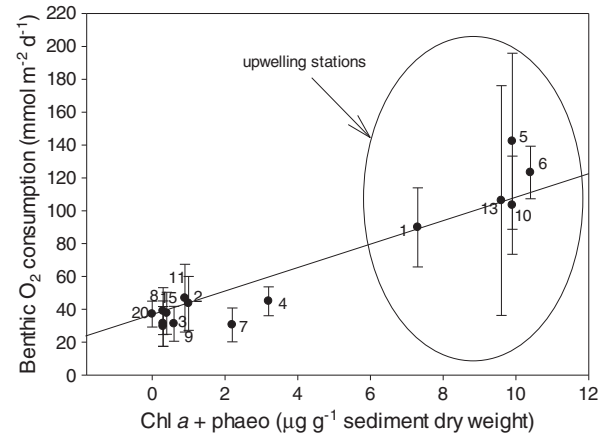
[28] Chlorophyll *a* and phaeopigment concentrations at the sediment surface (Table 2) varied from  $0.0$  to  $4.0 \mu\text{g g}^{-1}$  and from  $0.0$  to  $7.0 \mu\text{g g}^{-1}$  sediment dry weight, respectively. Both chlorophyll *a* ( $r = +0.873$ ;  $P < 0.0001$ ) and phaeopigment ( $r = +0.841$ ;  $P < 0.0001$ ) concentrations correlated linearly with water depth (Figure 5).

### 3.3. Surface Sediment Metabolism

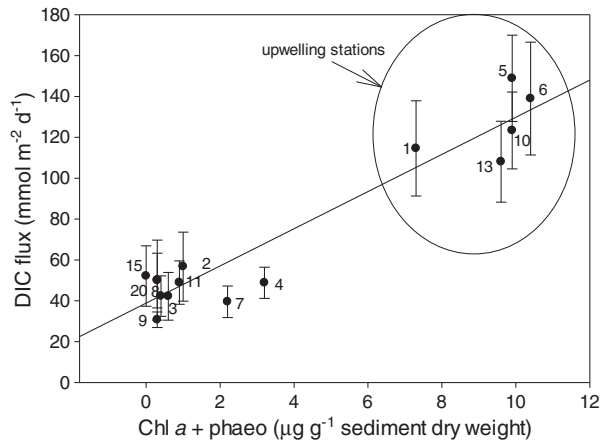
[29] Rates of oxygen consumption (Figure 6) ranged from  $29.6$  to  $142.3 \text{ mmol m}^{-2} \text{ day}^{-1}$ , and rates of DIC release (Figure 7) ranged from  $30.7$  to  $148.9 \text{ mmol m}^{-2} \text{ day}^{-1}$  among stations. Rates of both  $\text{O}_2$  uptake and DIC metabolism in



**Figure 5.** Surface sediment concentrations ( $\mu\text{g g}^{-1}$ ) of chlorophyll *a*, phaeopigment, and  $\delta^{15}\text{N}$  (‰) in relation to bottom water nitrate concentrations as a proxy indicator of upwelling. All regressions are significant ( $P < 0.001$ ).  $\text{Chla} = -0.676 + 0.162x$ ,  $R^2 = 0.797$ ,  $F_{1,18} = 66.682$ ;  $\text{phaeo} = -1.510 + 0.286x$ ,  $R^2 = 0.717$ ,  $F_{1,18} = 43.049$ ;  $\delta^{15}\text{N} = 2.179 + 0.048x$ ,  $R^2 = 0.516$ ,  $F_{1,18} = 18.102$ .



**Figure 6.** Rates of benthic oxygen consumption ( $\text{mmol m}^{-2} \text{ day}^{-1}$ ) versus total pigment concentrations ( $\mu\text{g g}^{-1}$ ) in surface sediments at each of the stations along the southern Timorese shelf. Stations are identified against each symbol. Values are mean  $\pm 1$  standard deviation. Regression:  $\text{O}_2 = 29.666 + 8.732x$ ,  $R^2 = 0.921$ ,  $F_{1,14} = 151.602$  ( $P < 0.001$ ).



**Figure 7.** Rates of benthic DIC release ( $\text{mmol m}^{-2} \text{ day}^{-1}$ ) across the sediment-water interface versus total pigment concentrations ( $\mu\text{g g}^{-1}$ ) in surface sediments at each of the stations along the southern Timorese shelf. Stations are identified against each symbol. Values are mean  $\pm 1$  standard deviation. Regression:  $\text{DIC} = 38.848 + 9.103x$ ,  $R^2 = 0.908$ ,  $F_{1,14} = 129.025$  ( $P < 0.001$ ).

surface sediments related best to surface concentrations of chlorophyll *a* + phaeopigment. Linear regression coefficients of plant pigments with  $\text{O}_2$  and DIC metabolism were  $r = +0.939$  and  $r = +0.921$ , respectively, with highest metabolic rates at Stas. TLS1, TLS5, TLS6, TLS10, and TLS13 (Figures 6 and 7). Both metabolic measurements did not correlate significantly with any other sediment property.

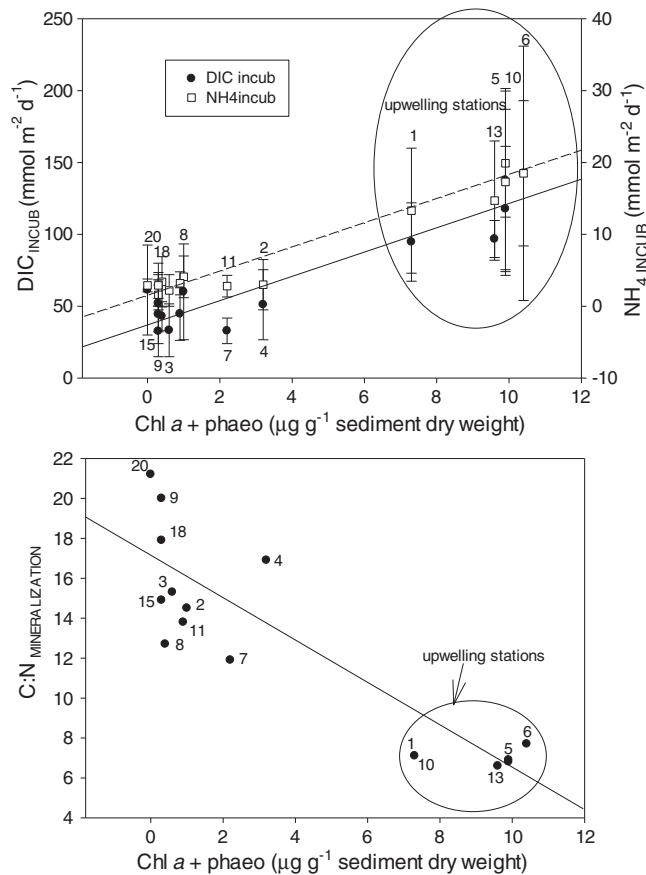
### 3.4. DIC and $\text{NH}_4^+$ Production

[30] Rates of DIC production in incubated sediments agreed well with DIC release across the sediment-water interface, with rates ranging from  $32.6$  to  $142.5 \text{ mmol m}^{-2} \text{ day}^{-1}$

(Figure 8, top). Rates of  $\text{NH}_4^+$  production mirrored DIC production rates in correlating significantly with surface concentrations of chlorophyll *a* + phaeopigment (Figure 8, top). The molar ratios of mineralized DIC: $\text{NH}_4^+$  correlated inversely with total pigment concentrations (Figure 8, bottom).

### 3.5. Rates of Sulfate, Iron, and Manganese Reduction

[31] Sulfate reduction dominated at stations toward the eastern end of the island, Stas. TLS8, TLS9, TLS15, TLS18, and TLS20 (Table 3). Sulfate reduction correlated inversely with  $\delta^{15}\text{N}$  ( $r = -0.713$ ;  $P < 0.00283$ ) and positively with percentage terrigenous TOC ( $r = +0.714$ ;  $P < 0.0028$ ) and percentage TOC ( $r = +0.673$ ;  $P < 0.006$ ). Iron reduction did not correlate significantly with any other sediment property. Manganese reduction dominated anaerobic microbial metabolism at 9 of the 15 stations (Table 3), but iron reduction was dominant only at Sta. TLS11. Manganese reduction correlated with water depth

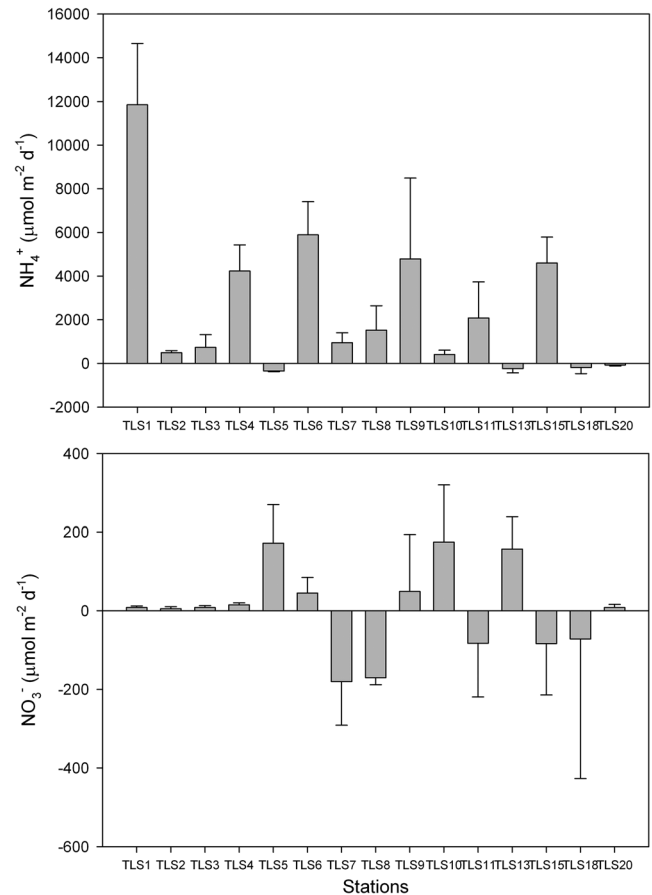


**Figure 8.** (top) Rates of benthic DIC and  $\text{NH}_4^+$  production ( $\text{mmol m}^{-2} \text{day}^{-1}$ ), and (bottom) molar ratios of DIC: $\text{NH}_4^+$  release from incubated sediments versus total pigment concentrations in surface sediments at each of the stations along the southern Timorese shelf. Stations are identified against each symbol from incubated sediments. Values are mean  $\pm 1$  standard deviation. Regressions:  $\text{DIC}_{\text{INC}} = 38.561 + 8.284x$ ,  $R^2 = 0.860$ ,  $F_{1,14} = 79.912$ ;  $\text{NH}_{4\text{inc}} = 1.609 + 1.575x$ ,  $R^2 = 0.946$ ,  $F_{1,14} = 229.637$ ;  $\text{C:N}_{\text{min}} = 16.769 - 1.018x$ ,  $R^2 = 0.763$ ,  $F_{1,14} = 41.844$ . All regressions are significant ( $P < 0.001$ ).

**Table 3.** Mean ( $\pm 1$  standard deviation) Rates of Sulfate (SRR), Manganese (MnR), and Iron (FeR) Reduction in Surface (0–10 cm) Sediments at Stations Along the South Timorese Coast<sup>a</sup>

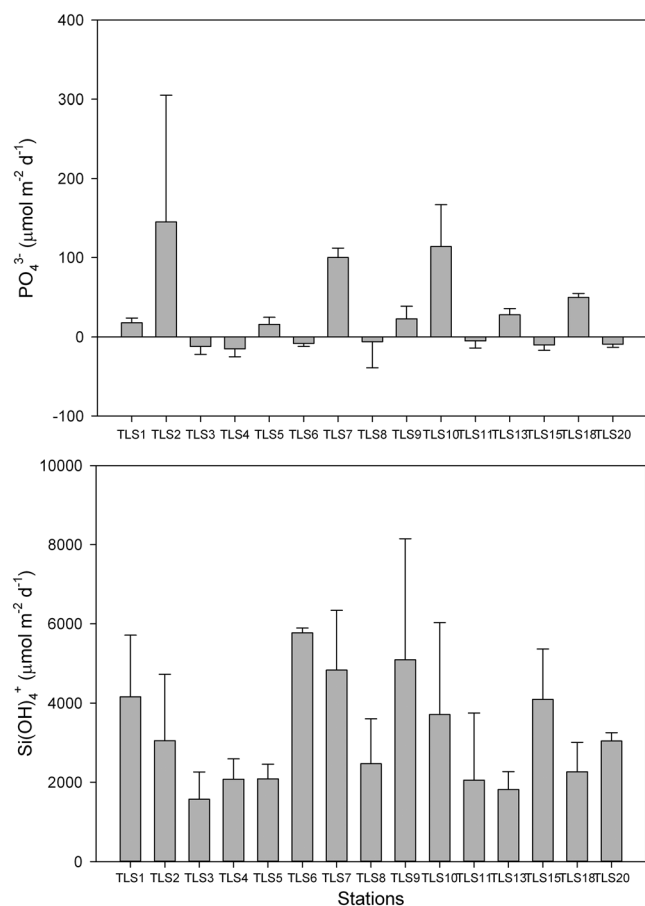
Station	SRR	MnR	FeR
TLS1	3.8 $\pm$ 1.4 (4%)	20.2 $\pm$ 3.4 (21%)	2.1 $\pm$ 2.2 (2%)
TLS2	2.0 $\pm$ 0.4 (3%)	18.7 $\pm$ 4.0 (31%)	0.0 $\pm$ 0.0
TLS3	2.2 $\pm$ 0.6 (7%)	3.2 $\pm$ 1.8 (10%)	0.0 $\pm$ 0.0
TLS4	4.8 $\pm$ 1.4 (9%)	7.0 $\pm$ 0.0 (14%)	1.1 $\pm$ 0.8 (2%)
TLS5	2.4 $\pm$ 0.4 (2%)	25.4 $\pm$ 6.1 (18%)	1.8 $\pm$ 1.9 (1%)
TLS6	8.8 $\pm$ 2.0 (6%)	35.3 $\pm$ 2.1 (25%)	6.0 $\pm$ 4.1 (4%)
TLS7	1.8 $\pm$ 0.6 (5%)	2.2 $\pm$ 2.0 (7%)	0.0 $\pm$ 0.0
TLS8	20.4 $\pm$ 7.6 (47%)	2.4 $\pm$ 1.8 (6%)	6.1 $\pm$ 4.1 (14%)
TLS9	7.2 $\pm$ 2.0 (21%)	1.1 $\pm$ 1.4 (3%)	1.0 $\pm$ 1.0 (3%)
TLS10	1.0 $\pm$ 0.2 (<1%)	34.5 $\pm$ 3.1 (29%)	1.1 $\pm$ 2.0 (1%)
TLS11	0.6 $\pm$ 0.2 (1%)	0.0 $\pm$ 0.0	3.0 $\pm$ 3.0 (7%)
TLS13	18.4 $\pm$ 2.2 (19%)	24.6 $\pm$ 3.4 (25%)	0.0 $\pm$ 0.0
TLS14	n.d.	n.d.	n.d.
TLS15	13.2 $\pm$ 1.8 (30%)	0.0 $\pm$ 0.0	0.0 $\pm$ 0.0
TLS16	n.d.	n.d.	n.d.
TLS17	n.d.	n.d.	n.d.
TLS18	3.8 $\pm$ 0.8 (7%)	0.0 $\pm$ 0.0	0.0 $\pm$ 0.0
TLS19	n.d.	n.d.	n.d.
TLS20	37.6 $\pm$ 10.2 (61%)	0.0 $\pm$ 0.0	0.0 $\pm$ 0.0

<sup>a</sup>Rates are in carbon equivalents ( $\text{mmol C m}^{-2} \text{day}^{-1}$ ) based on stoichiometric equations in Alongi [1998]. Values in parentheses are the percentage contribution of the specific pathway to TC metabolism in incubated sediments ( $\text{DIC}_{\text{INCUB}}$  in Figure 8, top panel). n.d., not detected.



**Figure 9.** Rates of (top)  $\text{NH}_4^+$  and (bottom)  $\text{NO}_3^-$  flux ( $\mu\text{mol m}^{-2} \text{day}^{-1}$ ) across the sediment-water interface at each of the stations along the southern Timorese shelf. Values are mean  $\pm 1$  standard deviation.





**Figure 10.** Rates of (top)  $\text{PO}_4^{3-}$  and (bottom)  $\text{Si}(\text{OH})_4^+$  flux ( $\text{mmol m}^{-2} \text{day}^{-1}$ ) across the sediment-water interface at each of the stations along the southern Timorese shelf. Values are mean  $\pm 1$  standard deviation.

( $r=+0.792$ ;  $P < 0.0004$ ),  $^{15}\delta \text{N}$  ( $r=+0.589$ ;  $P < 0.02$ ), bottom water nitrate ( $r=+0.866$ ;  $P < 0.00002$ ), total pigments, chlorophyll *a* and phaeopigments ( $r$  range =  $+0.84$  to  $+0.93$ ), and  $\text{NO}_3$  flux ( $r=+0.724$ ;  $P < 0.00228$ ).

### 3.6. Nutrient Fluxes Across the Sediment-Water Interface

[32] Patterns of uptake and release of  $\text{NH}_4^+$  (Figure 9, top),  $\text{NO}_3^-$  (Figure 9, bottom),  $\text{PO}_4^{3-}$  (Figure 10, top), and  $\text{Si}(\text{OH})_4^+$  (Figure 10, bottom) across the sediment-water interface were highly variable among stations and replicate chambers.  $\text{NH}_4^+$ ,  $\text{PO}_4^{3-}$ , and  $\text{Si}(\text{OH})_4^+$  fluxes did not correlate with any variables, but  $\text{NO}_3^-$  fluxes did correlate with chlorophyll *a* ( $r=+0.691$ ;  $P < 0.0043$ ), phaeopigments ( $r=+0.717$ ;  $P < 0.0026$ ), oxygen consumption ( $r=+0.728$ ;  $P < 0.0021$ ), and  $\text{NH}_4^+$  production in incubated sediments ( $r=+0.716$ ;  $P < 0.00267$ ).

## 4. Discussion

### 4.1. Intrusions of ITF Water

[33] Vertical water-column structure, bottom water nitrate and phosphate concentrations, and the OceanMAPS model results confirm that ITF intrusions take place along the entire southern margin of Timor, at least during the dry winter monsoon period. Most shelf waters, especially west of

longitude  $126^\circ 25'$  (Sta. TLS13) were either highly stratified or partially stratified, with a clearly defined surface mixed layer. At the eastern end of Timor near the Timor Passage, in situ temperature profiles (Figure 2) and model results suggest upwelling of submixed layer water to the ocean surface, possibly as an eddy induced by current instabilities and strong velocity shear [Talley *et al.*, 2011]. According to previous studies [Molcard *et al.*, 1996; Condie, 2011], it is at the southeastern corner of the island that the ITF takes a more south-southwesterly direction and meets warmer surface waters sourced from the Banda Sea to the east.

[34] Along the shelf west of longitude  $126^\circ 25'$ , the ITF intrudes as a subsurface wedge beneath lower-salinity water (Figure 2, see Sta. TLS1 profile; Figure 3, middle, section 125 E). Salinities of surface waters along the entire shelf vary little between 33.6 and 33.8, reflecting both the residual effects of river runoff during and immediately after the summer wet season, and the fact that surface waters originate from the Banda Sea. A degree of mixing and dilution is evidenced by the gradual lineal decline in dissolved nutrient concentrations from the shelf break to the nearshore zone (Figure 4). The high  $\text{O}_2$  levels ( $>7 \text{ mg L}^{-1}$ ) and a salinity range of 34.3–34.6 below the top of the thermocline are consistent with the characteristics of the ITF being dominated by low-salinity, well-ventilated, upper thermocline North Pacific water [Molcard *et al.*, 1996; Sprintall, 2009; Schiller *et al.*, 2010].

[35] The vertical structure of Timorese shelf waters at other times of the year is unknown, but upwelling is likeliest to occur during the winter SE monsoon, as it does in the eastern Banda Sea [Gieskes *et al.*, 1990] and along the south coasts of Java and Sumatra [Susanto *et al.*, 2001]. With persistent SE trade winds blowing onshore, Ekman dynamics in the Southern Hemisphere means that shelf waters are displaced offshore in the surface layer, to be balanced by onshore flows in the lower part of the water column with an upwelling flow at the shelf-ocean boundary. Pulses of upwelling-favorable wind stress often correlate with onshore near-bed mean current flow [Simpson and Sharples, 2012], with concomitant declines in near-bed temperature indicating the on-shelf transfer of cooler, deeper water, and such appears to be the case along the Timorese coast. Fronts between cold, upwelled water and warmer, more dilute, shelf surface water are generally not stable [Johnson and Rock, 1986], with alongshore geostrophic currents on the warm side of the front generating mesoscale frontal instabilities and possible eddies. A number of small eddies, no larger than  $\approx 70$ – $100 \text{ m}$  in diameter, were observed visually, frequently in proximity to river plumes, and these phenomena may be accentuated by strong tidal forces. Along the shelf east of longitude  $126^\circ 25'$ , large areas of the seabed were sandy and appeared to be scoured, possibly by these onshore near-bed currents induced by wind stress and strong tides (maximum tidal range  $> 3 \text{ m}$ ) [Egbert *et al.*, 1994]. This would account for the lack of benthic enrichment along the far eastern section of the southern shelf.

[36] Intrusions of ITF water supply high concentrations of nitrate, phosphate, and silicate onto the shelf (Figure 4), whereas wind-driven offshore surface flow and other mechanisms such as eddies may export surface material off the shelf. The high concentrations of dissolved nutrients and the sharp vertical gradients in oxygen, temperature,

and salinity likely stimulated phytoplankton productivity. We were unprepared to measure primary production and chlorophyll biomass in the water column, so the chlorophyll *a* profiles measured via the calibrated fluorometer must be considered with caution and in relative terms. Our fluorescence values were exceedingly high ( $>5 \text{ mg m}^{-3}$ ), greater than measured chlorophyll levels ( $\leq 1.5 \text{ mg m}^{-3}$ ) within the upwelling zones off Java and Sumatra [Susanto and Marra, 2005], and off the northwest Australian shelf [Furnas, 2007; McKinnon *et al.*, 2011], so they are unlikely to be correct. The high fluorescence levels likely reflect high levels of dissolved and particulate matter originating from river runoff. However, the vertical trends in fluorescence do suggest that peak plankton biomass and productivity occurred just above the 1% isolume and the thermocline at many sites (Figure 2). The comparatively high concentrations of chlorophyll and phaeopigment in surface sediments beneath the areas of subsurface upwelling offer further support for the idea of high primary productivity, as the sediment pigments must be the net result of deposition to the seabed from the overlying water column; benthic in situ production is highly unlikely given the shallow depths of the 1% light level along the shelf.

[37] Surface upwelling off the eastern end of Timor-Leste does not seem to have enhanced concentrations of nitrate, phosphate, and chlorophyll inshore at Stas. TLS19 and TLS20 (Figures 4 and 5). The CTD casts at these sites suggest that upwelling at the island's edge did not intrude close inshore, or at least may have been well mixed and diluted by strong surface flow of warmer water. The Ocean-MAPs results cannot be used to support the CTD data that upwelling occurs right up to the coast because model resolution was too coarse (10 km) to accurately delimit spatial patterns so close inshore.

[38] The Timor situation is analogous to the pattern of upwelling along the west coast of New Caledonia [Hénin and Cresswell, 2005] where coastal upwelling follows strong trade wind episodes that develop mostly in summer as a band of cold water offshore of the barrier reef; upwelling is the dominant process at daily time scales, and sea surface temperature is strongly influenced by seasonal variations in subsurface stratification [Alory *et al.*, 2006]. The biological response to upwelling events is localized enhancement of offshore plankton activity followed by calm periods when there is a noticeable relaxation of plankton growth and production [Ganachaud *et al.*, 2010].

#### 4.2. Benthic Response

[39] The distribution of fine sediments on the shelf delineates the long-term net deposition of silt and clay-sized particles derived from both marine and terrestrial sources. Most deposits of smallest grain size on the Timorese shelf are located west of longitude  $126^{\circ}25'$ , reflecting the preponderance of rivers, steep catchments, and higher rainfall mid-island compared to the eastern end of the island. The sediment  $^{15}\delta\text{N}$  values related positively to concentrations of dissolved nutrients in bottom waters and to both chlorophyll *a* and phaeopigment concentrations, suggesting enrichment of surface sediments in relation to enhanced primary producers associated with upwelling.

[40] The  $^{13}\delta\text{C}$  values suggest a more complex picture, with the percentage of terrestrial organic carbon (Table 2) varying from a maximum of 100% off the Motaarapomaco River (Sta. TLS20) where we found extensive patches of litter derived mostly from *Lontar* palm overlying sand deposits to a minimum of 22% at one of the western stations (Sta. TLS13). Some sites (e.g., Stas. TLS7 and TLS11) also have sedimentary organic carbon that is mostly marine derived, reflecting the fact that complex distribution and transport processes must operate on this shelf. Nevertheless, on average, the western areas of subsurface upwelling had proportionally more marine-derived organic carbon (63%) than the stations (40%) further east of longitude  $126^{\circ}25'$ .

[41] Benthic metabolism was greatly enhanced beneath the subsurface intrusions of the ITF. Rates of oxygen consumption and DIC release across the sediment-water interface, and rates of DIC and ammonium production within incubated sediments correlated significantly with benthic concentrations of chlorophyll *a* and phaeopigment. Benthic metabolic activities were clearly stimulated by the deposition of phytodetritus and fresh plankton debris as a direct result of phytoplankton production enhanced by upwelling of nutrient-rich water at depth [Asanuma *et al.*, 2003]. Benthic communities, especially microbial assemblages, have long been known to be enriched by rapid inputs of high-quality organic matter [Hanson *et al.*, 1981]. This scenario is identical to close benthic-pelagic coupling observed in other coastal upwelling ecosystems [Thiel, 1978; Barber and Smith, 1981]. What is unusual in the Timor shelf upwelling zones are the very high rates of benthic metabolism. These rates are among the highest recorded from marine sediments [Middelburg *et al.*, 2005], including from other tropical shelves where rates of oxygen consumption are usually less than  $50 \text{ mmol m}^{-2} \text{ day}^{-1}$  [Alongi, 1995, Alongi *et al.*, 2007; Aller *et al.*, 1996, 2008]. They are also unusually high considering the relatively low bottom water temperatures and the low ( $<1\%$ ) sediment organic carbon content on this shelf.

[42] The sediments (Sta. TLS20) with the highest (2%) TOC content due to significant outwelling of vascular plant detritus did not have especially rapid rates of metabolism (Figures 5 and 6) despite strong evidence of surface upwelling further offshore. As noted earlier, strong currents were observed in the region, possibly leading to scouring of the seabed rather than deposition of phytodetritus. The lower rates of benthic metabolism can also be explained by the fact that the palm tree debris observed at this site consisted of highly refractory organic matter, especially in comparison with plankton detritus, which is a more easily degradable and labile form of organic matter. Experiments have consistently shown that microbial metabolism is not readily stimulated by even high supply rates of vascular plant detritus, at least not without significant microbial enrichment and degradation over time [Hanson, 1982; Tenore *et al.*, 1982], and such appears to be the case at Sta. TLS20. A similar scenario exists in the Georgia Bight along the southeastern U.S. coast where the inner shelf experiences significant estuarine outwelling of salt marsh detritus, whereas the outer shelf and its phytoplankton communities are greatly stimulated by intrusions from the adjacent Gulf Stream. The pattern of benthic response off the south coast of Timor-Leste is similar in that rates of microbial metabolism were

significantly higher along the outer Timorese shelf. And just like the Georgia Bight, it is unlikely that high metabolic rates are sustained for long within the upwelling zones on the Timor shelf, as labile material is usually rapidly depleted within weeks of postbloom conditions.

[43] Our metabolic measurements were made only from surface sediments and very likely do not encompass the complete range of metabolic processes occurring in these shelf deposits. Measurable rates of microbial activity, especially iron reduction, have been observed to a sediment depth of at least 1 m in sediment cores taken off the Amazon and Fly rivers [Aller *et al.*, 1996, 2008], within the Great Barrier Reef [Alongi *et al.*, 2007], and off the rivers of southwest New Guinea [Alongi *et al.*, 2012]. Thus, the true rates and pathways of microbial metabolism on the Timor shelf are likely to be considerably different from the “snapshot” measurements taken during July 2011. Nevertheless, there are some clear patterns in the relative dominance of specific bacterial pathways in surface sediments. First, manganese reduction was rapid (Table 3) compared to rates measured in other shelf sediments [Aller, 1994; Thamdrup, 2000] and were closely linked to upwelling. Second, rates of iron reduction were low; significant iron reduction probably occurs in deeper sediment layers. Third, sulfate reduction dominated TC oxidation and was the only measurable anaerobic process in shelf sediments east of longitude 126°25'; correlation analysis indicates that sulfate reduction was linked not only to percentage organic carbon content, but also to the percentage of land-derived organic matter. Finally, although we measured only surface respiration, except for Stas. TLS8 and TLS20, 56% to 94% of TC decomposition is unaccounted for by the reduction of sulfate, iron, or manganese. Presuming that denitrification and methanogenesis were minor C oxidation pathways, which is not unreasonable considering the low organic carbon and nitrogen content of these deposits, by difference, aerobic respiration must play a significant role in surface sediments. This may not be the case if metabolism in deeper sediments had been measured, but in any case, oxic respiration plays a key role in driving benthic surface metabolism.

[44] In sediments beneath other upwelling zones, a suite of redox processes may dominate [Gallardo, 1977; Glud *et al.*, 1999; Sumida *et al.*, 2005]. For instance, under the Peruvian upwelling system, sulfate reduction and sulfide oxidation are major metabolic pathways in deep water sediments [Gallardo, 1977]. Regardless of the dominance of a specific diagenetic pathway, it is clear that upwelling events usually lead to enhanced benthic metabolism.

[45] Beneath the areas of subsurface upwelling, manganese reduction was the dominant metabolic pathway along with aerobic respiration. Sediment and near-bottom conditions apparently favored manganese-reducing bacteria; these assemblages outcompete other anaerobic microflora such as iron-reducers [Ehrlich and Newman, 2009]. Neither manganese- (Mn-) nor iron-reducing activities were measured in shelf sediments east of longitude 126°25' (Table 3) where upwelling was not observed close inshore. The study by Aller [1994] in Long Island Sound suggests that manganese-reducing bacteria can respond rapidly to deposition of phytodetritus, and such may be the case off Timor-Leste. Theoretically, diagenesis involving Mn can be linked to nitrate via coupled anaerobic nitrification/lithotrophic Mn reduction reactions [Ehrlich and Newman, 2009]. A significant

correlation ( $r=+0.724$ ) was found between Mn reduction and nitrate flux across the sediment-water interface, although this may reflect coincident, unrelated responses to the deposition and subsequent decomposition of phytopigments.

[46] Rates of sulfate reduction were low, equivalent to rates measured on other tropical shelves [Aller *et al.*, 1996, 2008; Alongi *et al.*, 2007, 2012], and correlated positively with concentrations of sediment TOC and the fraction of terrestrially derived OC, indicating a possible threshold of organic carbon concentration and source. Low rates of sulfate reduction in tropical sediments have been attributed to a number of factors, including frequent sediment disturbance and the predominance of highly weathered debris exported onto low-latitude shelves [Aller *et al.*, 1996, 2008; Alongi *et al.*, 2007, 2012]. These drivers may apply to sulfate-reducers in Timorese shelf deposits, but their low activities may reflect the fact that they are outcompeted by manganese-reducers [Ehrlich and Newman, 2009]. The dominance of aerobic respiration at many sites suggests conditions, such as high bottom water O<sub>2</sub> levels, unfavorable for the growth of sulfate-reducing bacteria. Like other tropical shelves, aerobic and suboxic diagenesis are the main pathways of organic matter decomposition off Timor-Leste.

[47] Seasonal upwelling off Timor-Leste may play an important role in supporting higher trophic levels in shelf and offshore food webs. Large, oceanic pelagic fish, such as the southern bluefin, albacore, bigeye, and yellowfin tuna, have been caught off the Timor shelf for the past 42,000 years [O'Connor *et al.*, 2011]. Tuna may be attracted to enhanced plankton productivity induced by upwelling at the shelf edge, as southern bluefin tuna are in the Great Australian Bight [Willis and Hobday, 2007]. The south coast of Timor is a key migration route for several species of whales and dolphins, and other megafauna (e.g., turtles) are frequently observed on the shelf [Butcher, 2004]. In the past, sperm whales were frequently hunted off the south coast of Timor [Beale, 1839; Butcher, 2004] and in upwelling areas of the adjacent Banda Sea [Bennett, 1840]. Whether large migratory megafauna are associated with the Timor upwelling is problematic but cannot be ruled out.

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